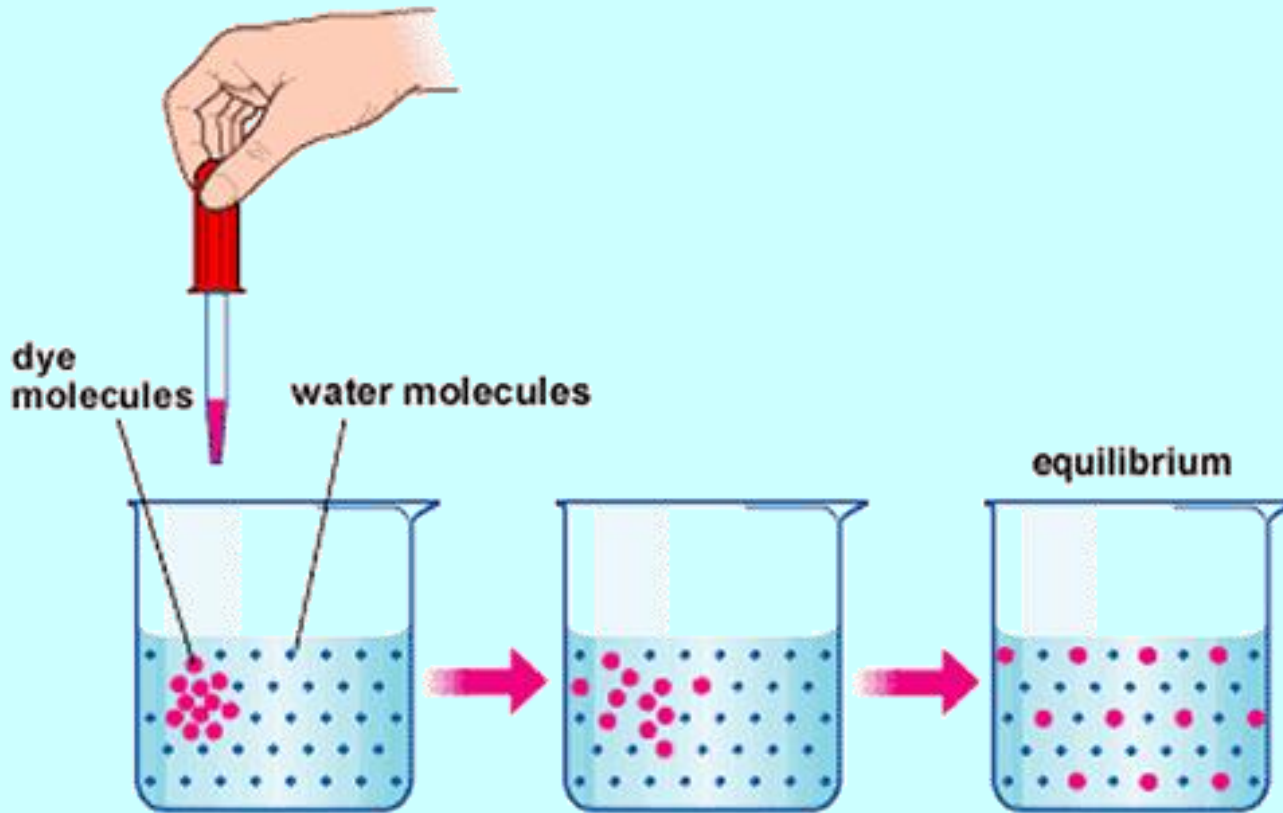
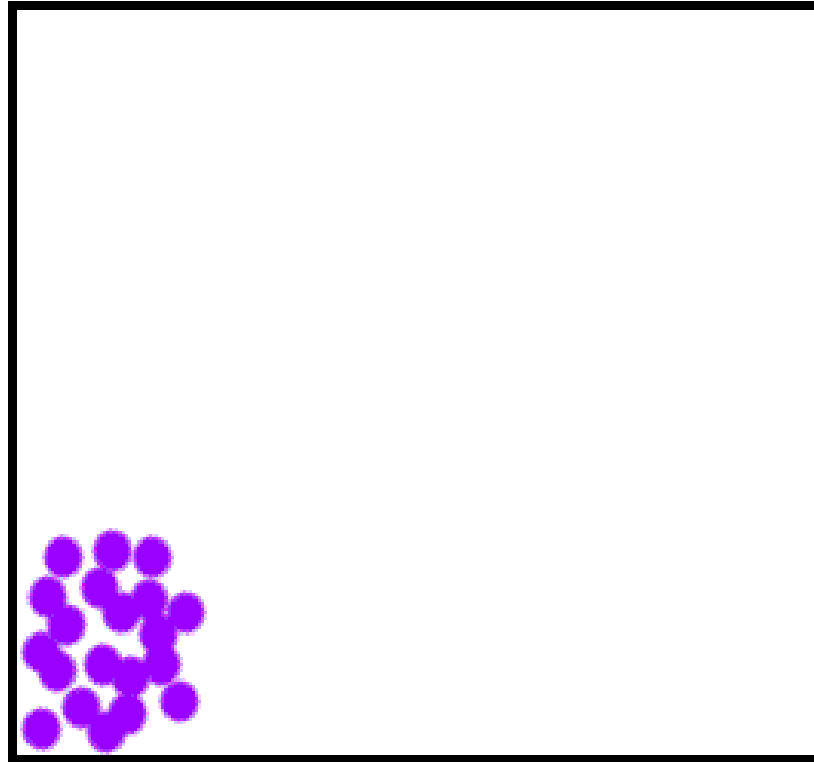


Diffusion





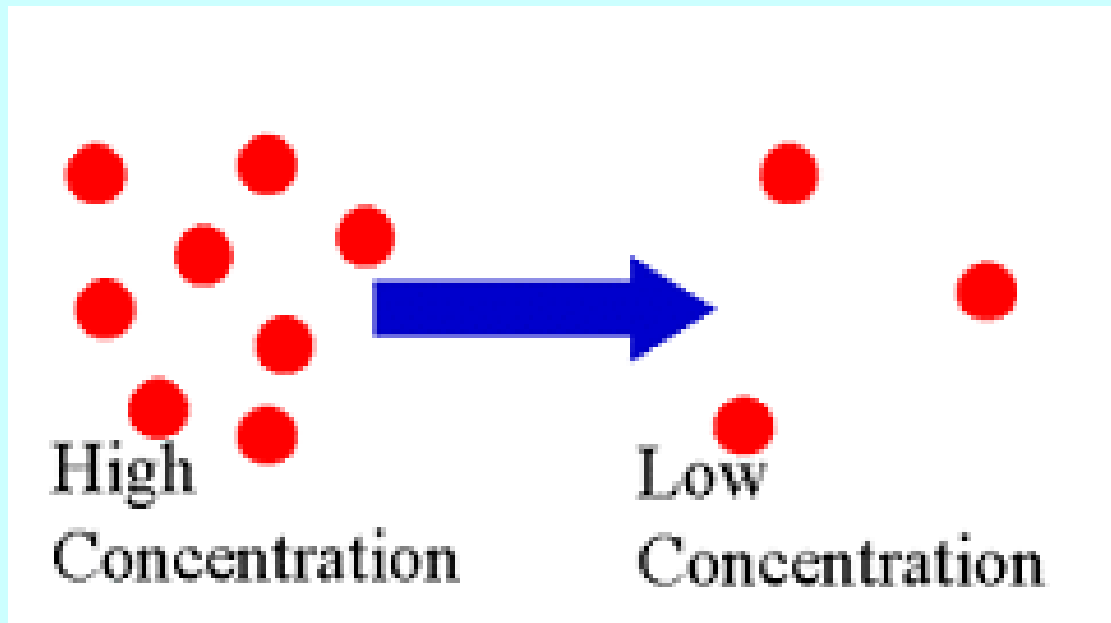
Molecules move

FROM "where there's A LOT"
to "where there's NOT"

DIFFUSION across a space

Happens anytime there is a DIFFERENCE in concentration in one place compared to another

= Concentration gradient



DIFFUSION across a SPACE

- Molecules move automatically DOWN the concentration gradient from an area of Higher concentration to an area of Lower concentration
- **EXAMPLES**

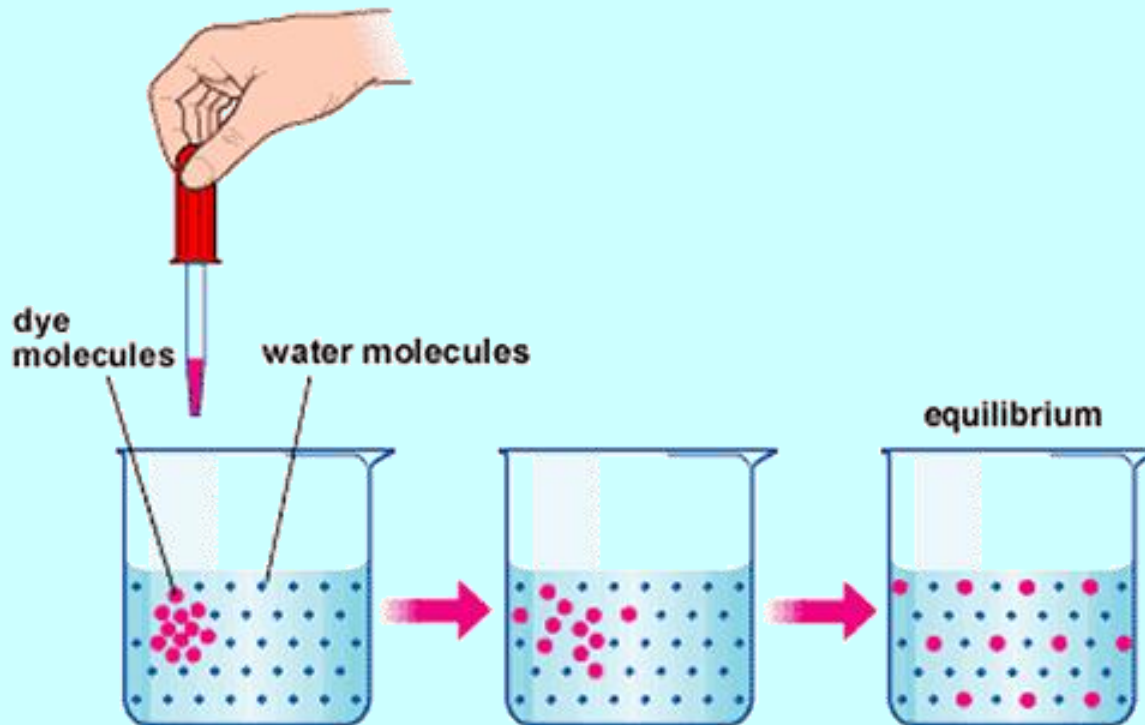
Blue dye in beaker demo,
Someone making popcorn/grilling out
Strong perfume,
Bad smell in room

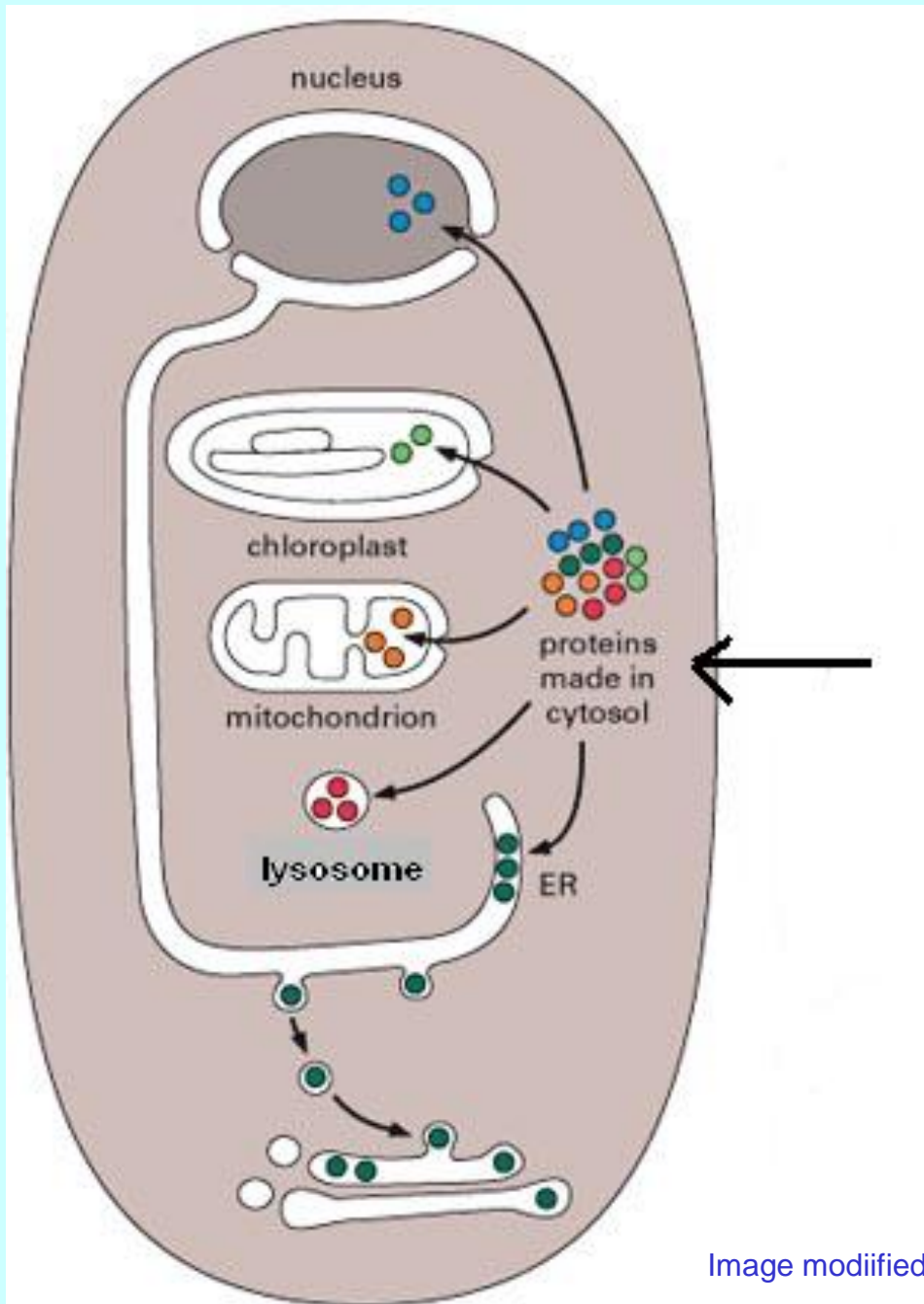


DIFFUSION across a space

Diffusion continues until the concentration is equal everywhere in space

= Equilibrium

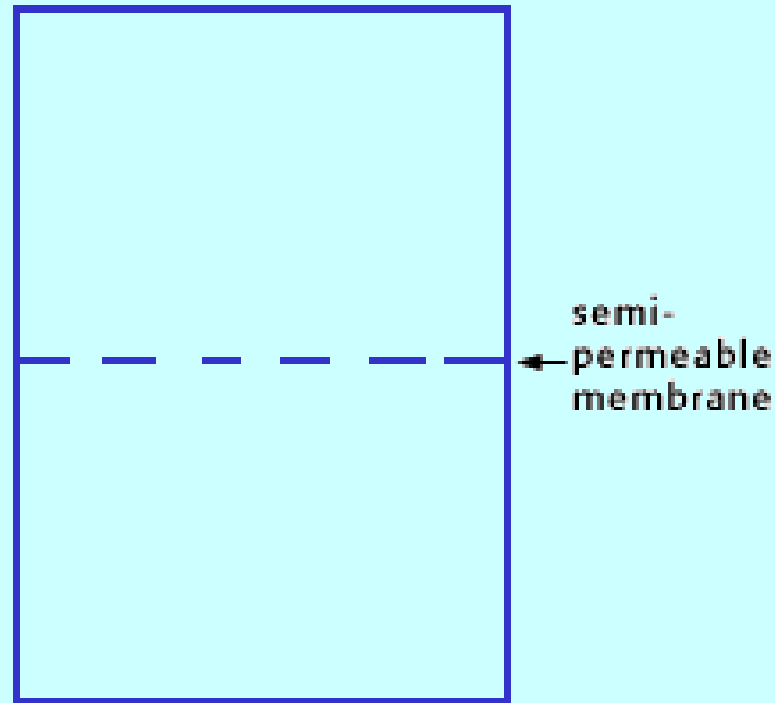




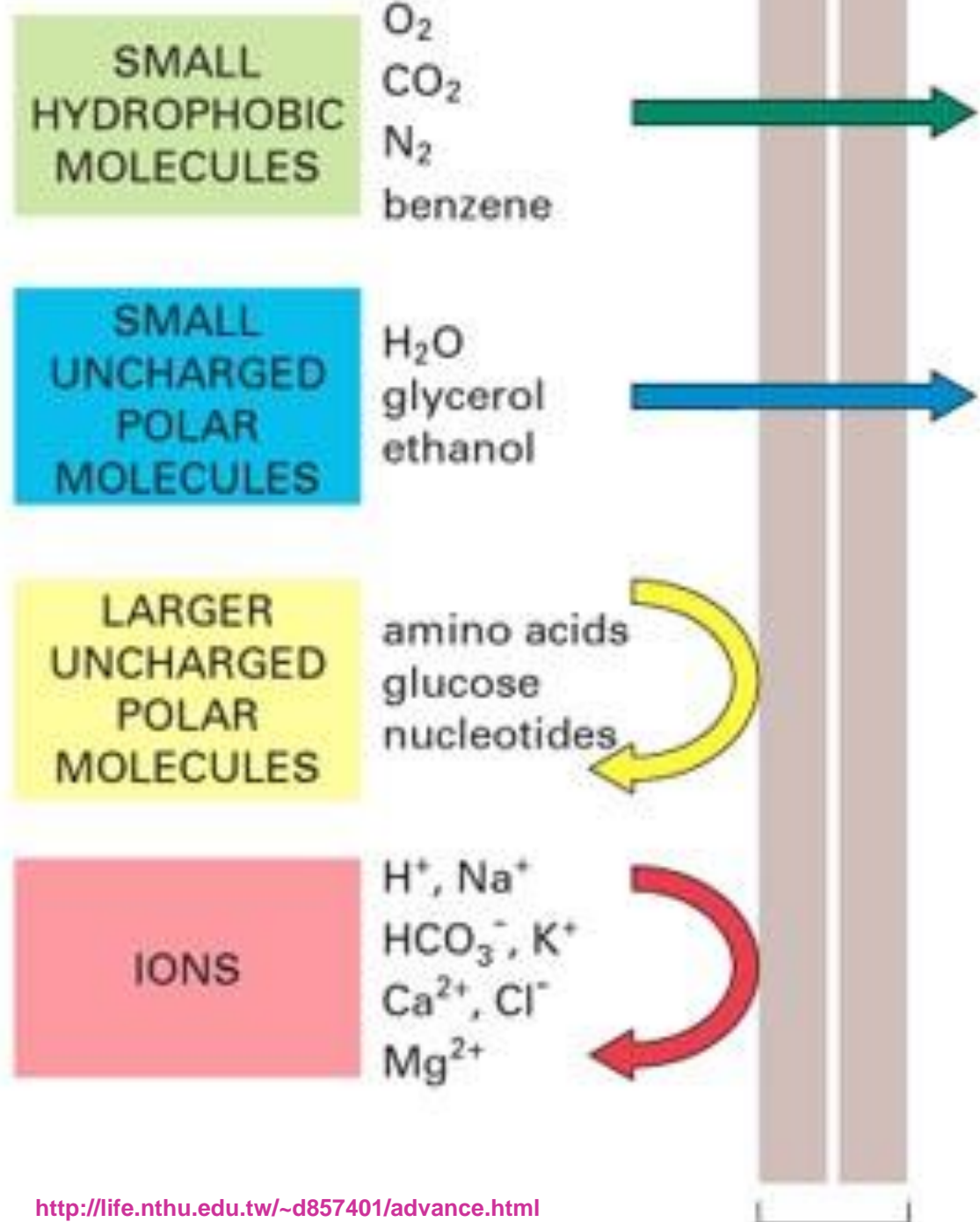
**Molecules
need to
move
across
membranes
in cells**

Diffusion can happen across a membrane in a cell, too

Diffusion across a membrane



...as long as membrane will let the molecule pass through



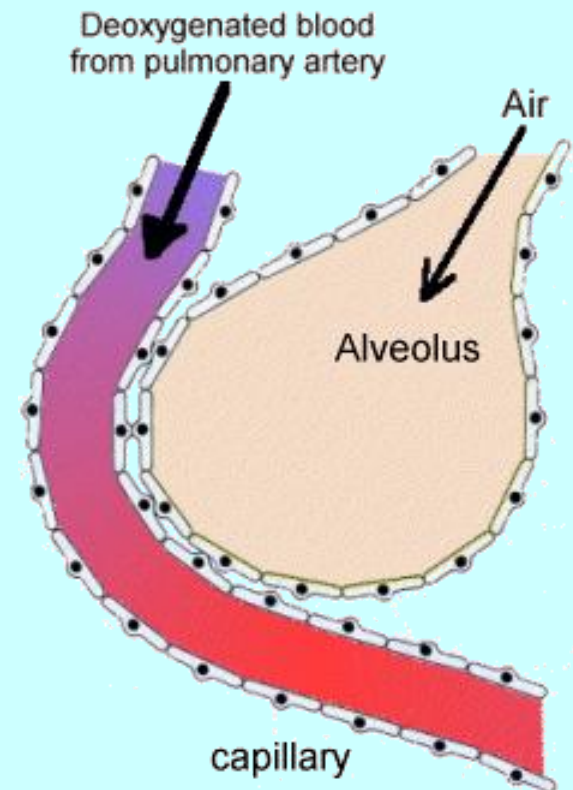
SELECTIVELY PERMEABLE (Semi-permeable)

[See a movie](#)

CELL EXAMPLE:

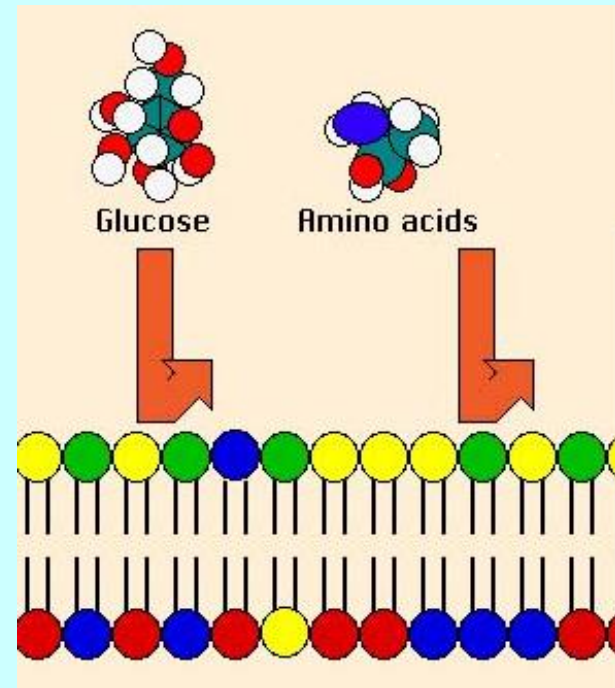
DIFFUSION automatically moves oxygen from **HIGHER** concentration (in lungs) to a **LOWER** concentration (in blood)

CO₂ automatically moves from where there is a **HIGHER** concentration (in blood) to where there is a lower concentration (in lungs)



BUT....

What if a cell needs to move LARGE molecules?

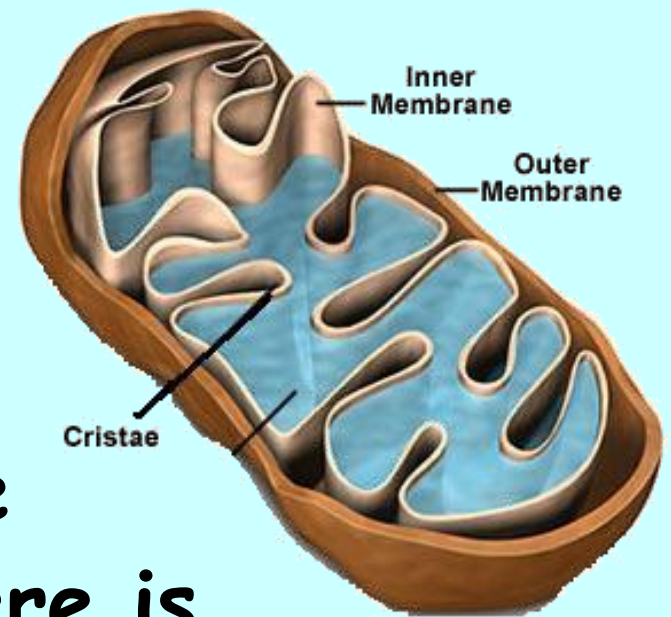


What if cell needs to move a molecule AGAINST the **CONCENTRATION GRADIENT?**

(LOWER → HIGHER)

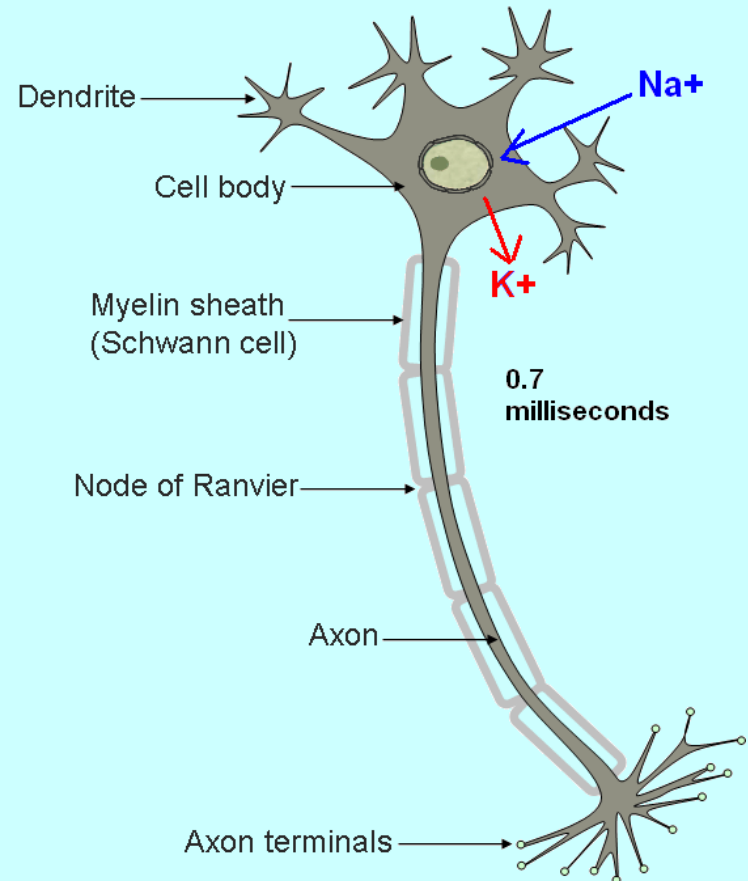
Cell example:

Want to put **MORE** glucose into mitochondria when there is already glucose in there



What if cell needs to move molecules really FAST ?
(can't wait for it to diffuse)

Cell example:
Movement of Na^+ & K^+ ions required to send nerve signals



We need a WAY to HELP
molecules across cell
membranes that can't go
across by themselves

KINDS OF TRANSPORT

PASSIVE (NO ENERGY required)

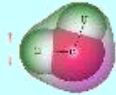
DIFFUSION

Oxygen

Carbon
Dioxide

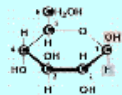
OSMOSIS

WATER



FACILITATED DIFFUSION

GLUCOSE



ION CHANNELS

Ca⁺⁺

H⁺

Na⁺

K⁺

ACTIVE (requires ENERGY)

Na⁺ - K⁺ PUMP

Na⁺

K⁺

ENDOCYTOSIS

Phagocytosis

Large molecules
Whole cells

Pinocytosis

Small molecules
Fluids

EXOCYTOSIS

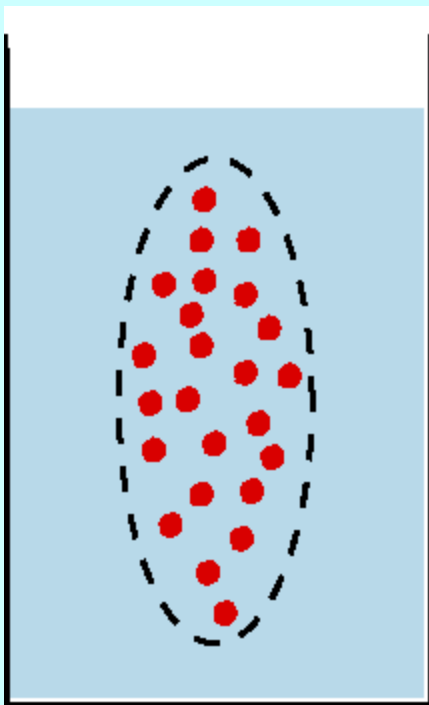
Kinds of **PASSIVE** Transport

- Diffusion

- Osmosis

DIFFUSION across a membrane

Happens anytime there is a DIFFERENCE in concentration on one side of the membrane compared to the other



Molecules that move by diffusion across membranes in cells:

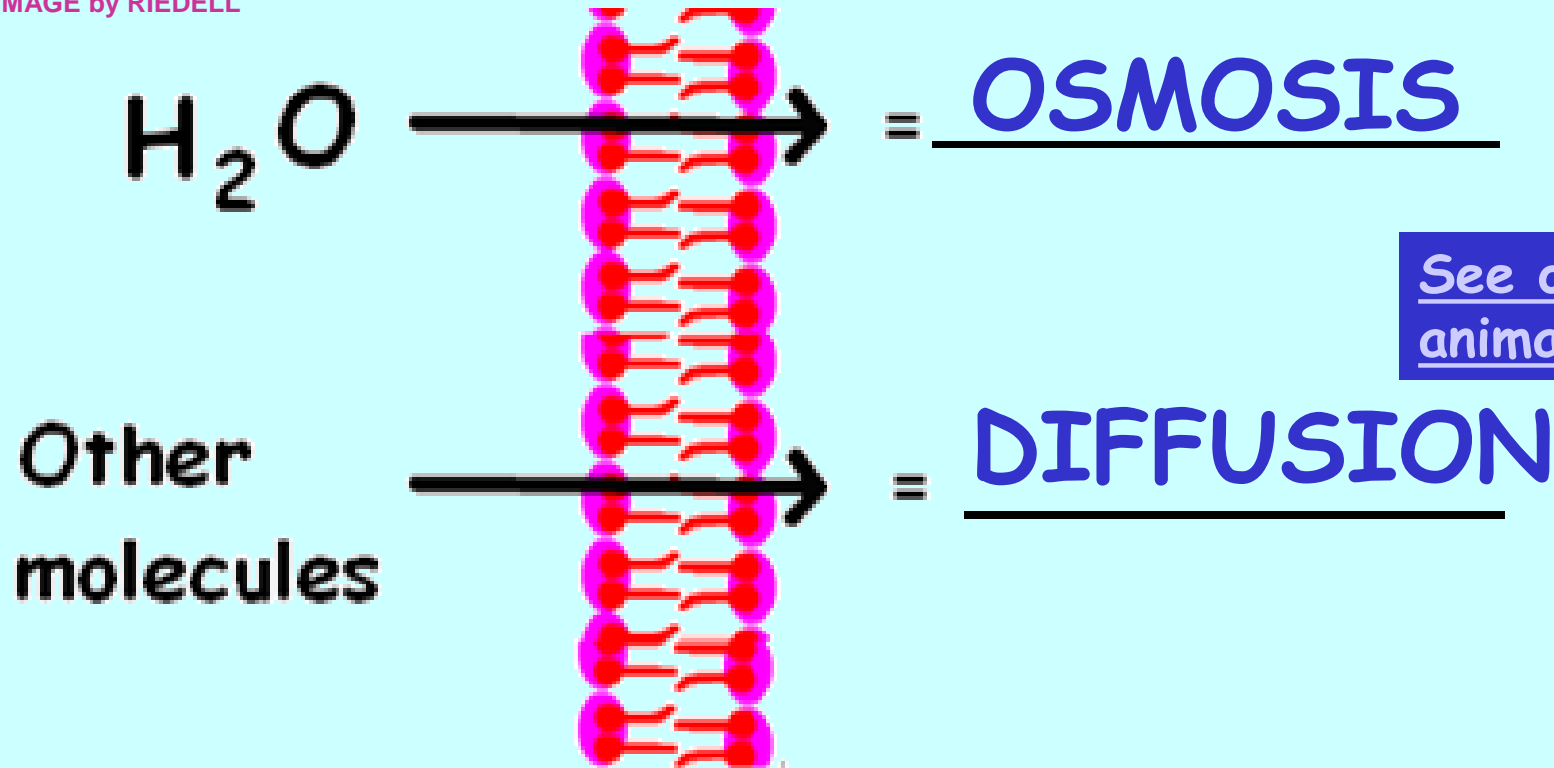
OXYGEN

CARBON DIOXIDE

OSMOSIS

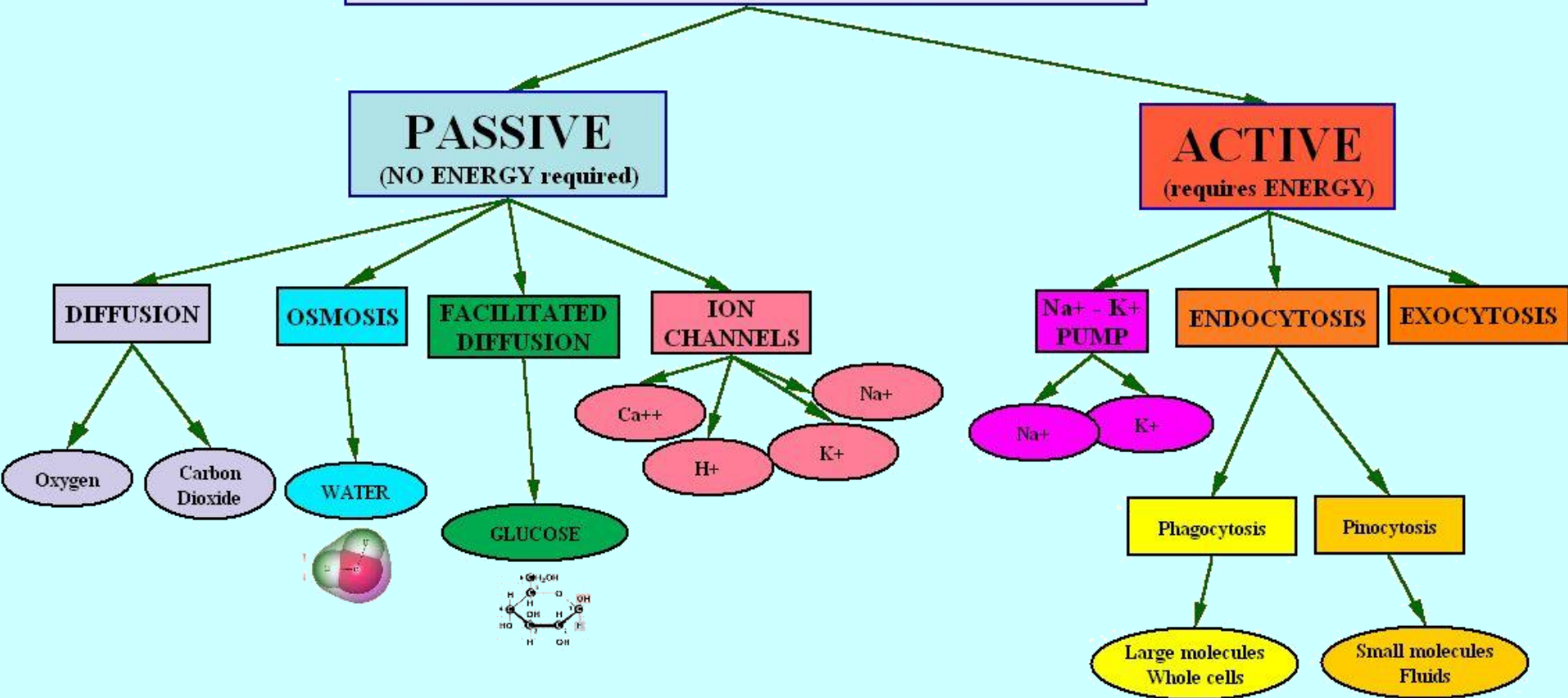
= SPECIAL KIND OF DIFFUSION

IMAGE by RIEDELL



Movement of molecules across a Semi-permeable membrane from Higher concentration to lower

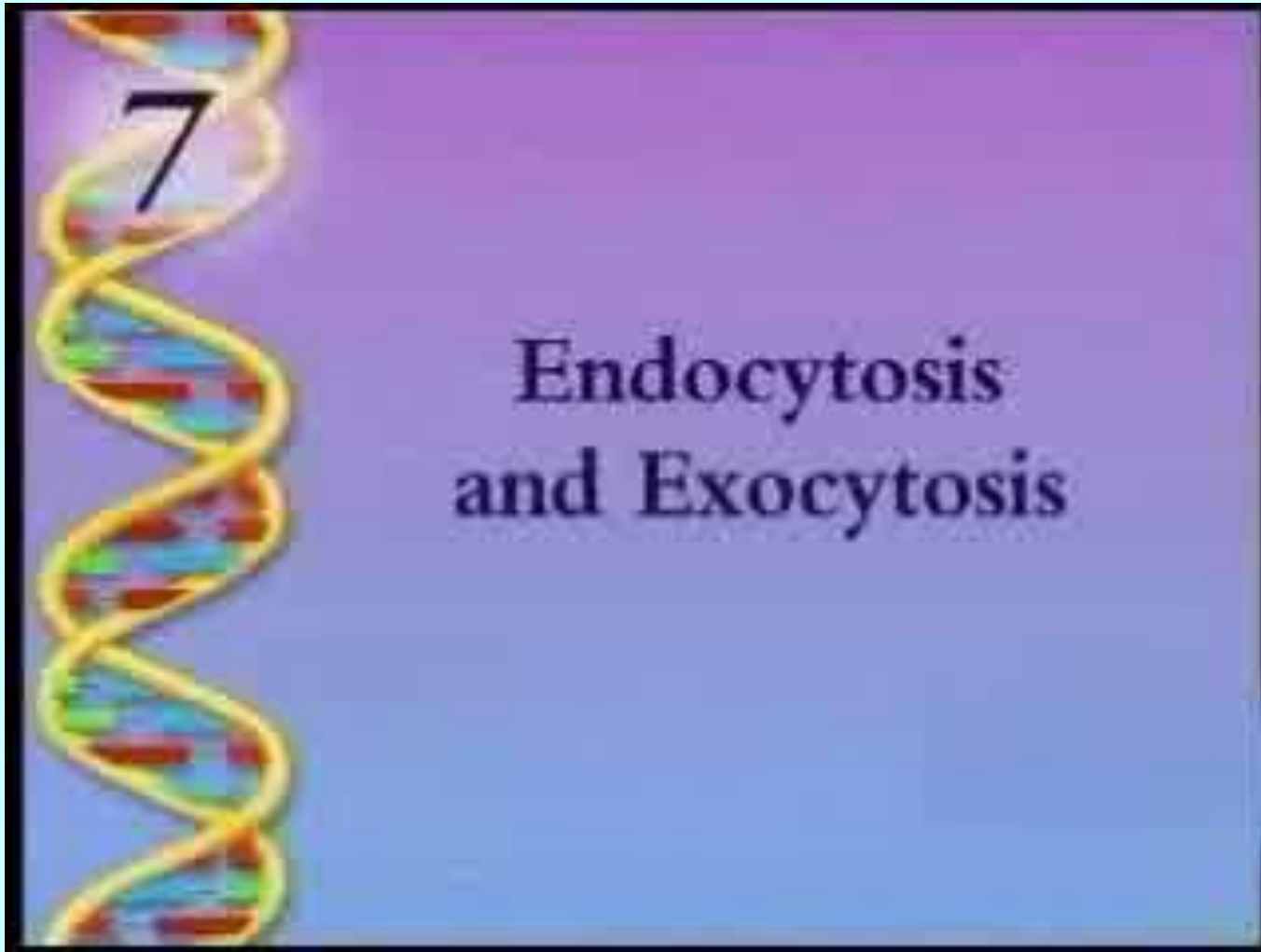
KINDS OF TRANSPORT



Kinds of ACTIVE Transport

- Endocytosis

- Exocytosis



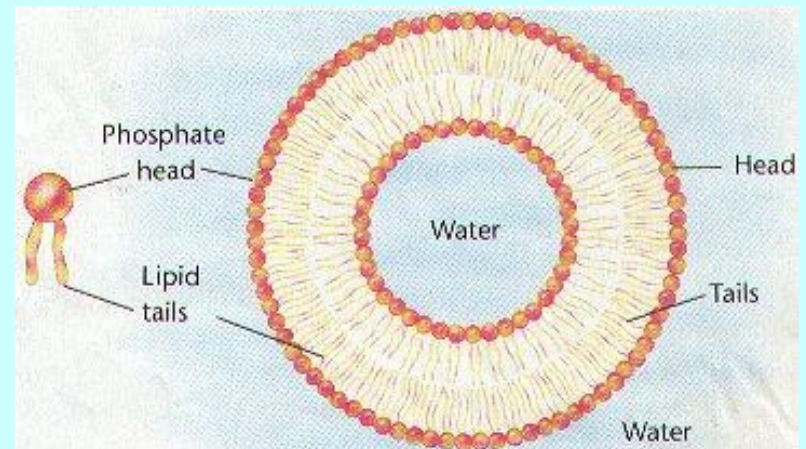
**See a video clip about
Endo/exocytosis -7E**



ENDOCYTOSIS

Takes substances into cell

- ACTIVE transport
(requires energy from ATP)
- Uses small membrane sacs called VESICLES to carry substances



2 KINDS of ENDOCYTOSIS

Takes substances into cell

If taking in:

fluid or small molecules = PINOCYTOSIS

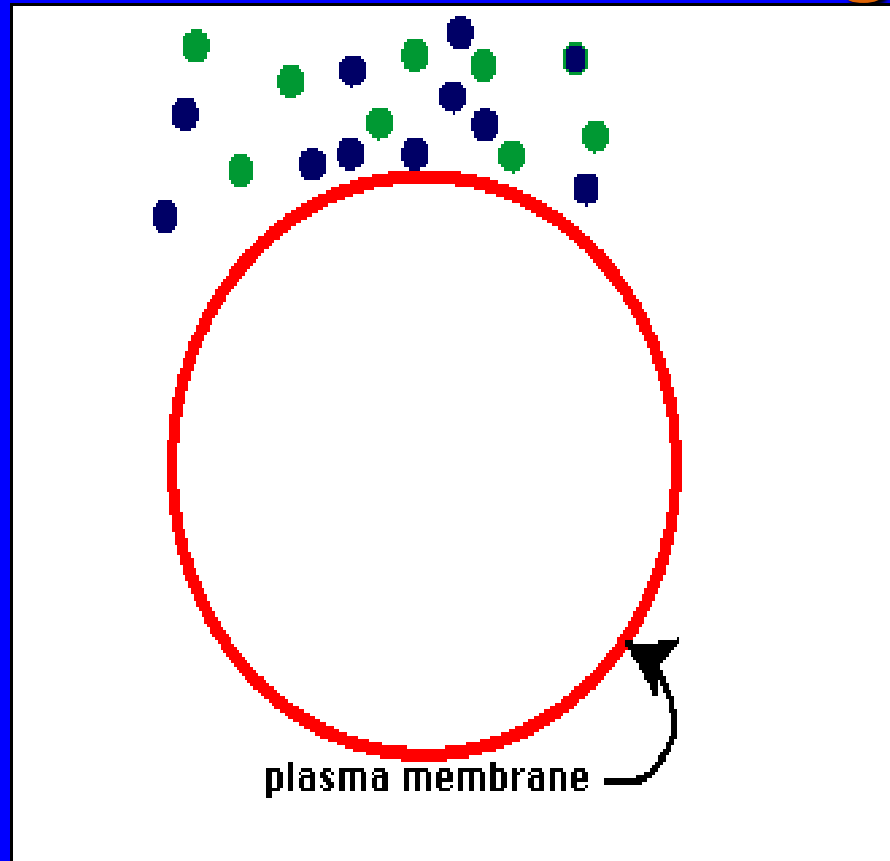
large particles or whole cells = PHAGOCYTOSIS

• Examples in cells:

- one celled organisms eat this way
- white blood cells get rid of bacteria this way

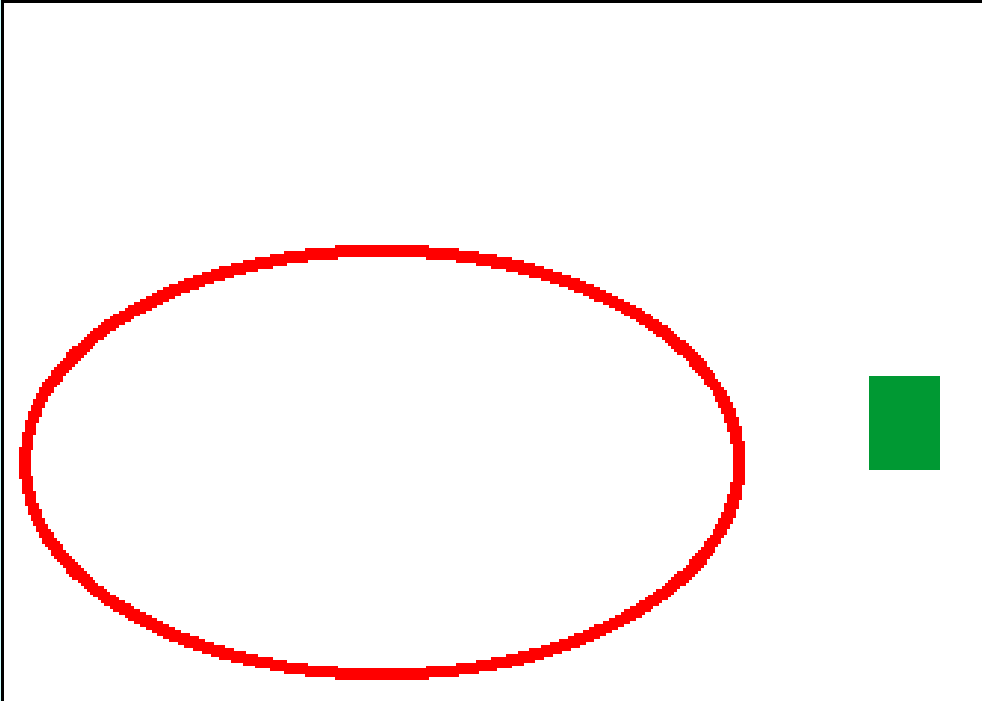
Pinocytosis

- Called "Cell Drinking"

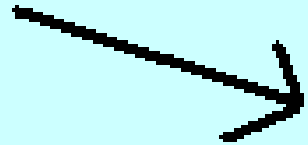
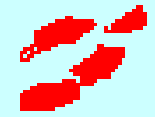
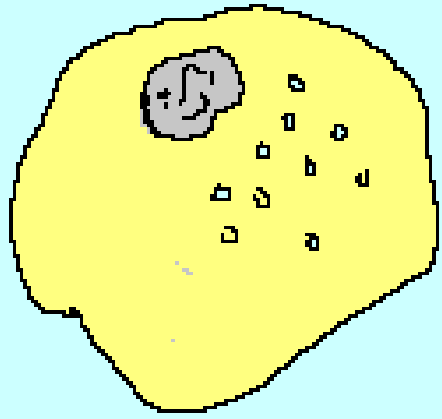


ENDOCYTOSIS

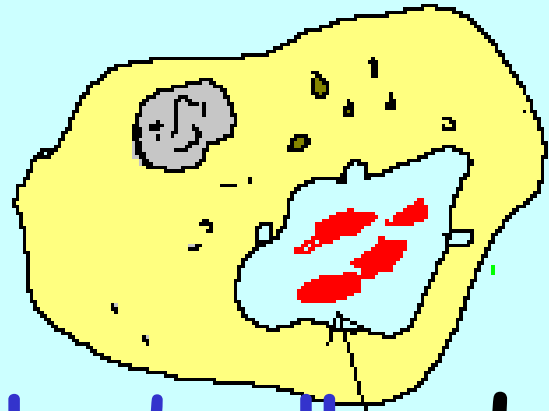
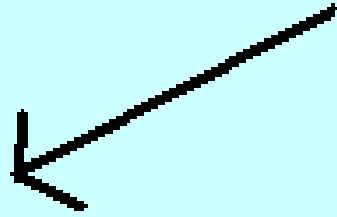
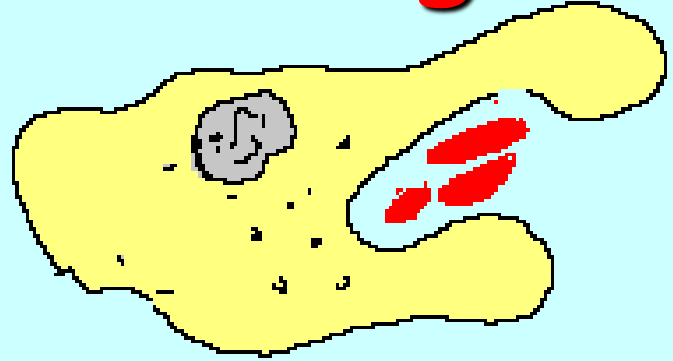
Animation from: <http://academic.brooklyn.cuny.edu/biology/bio4fv/page/cell-movement.html>



PHAGOCYTOSIS



Called "Cell Eating"



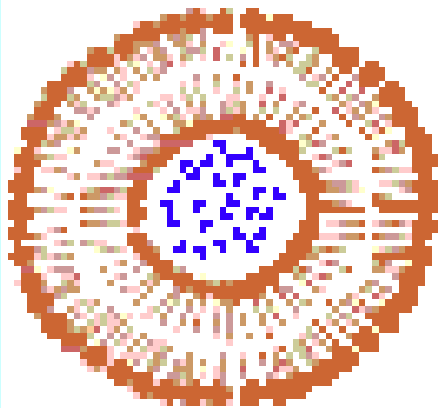
White blood cell destroying germs

EXOCYTOSIS

Substances released outside of cell

- ACTIVE transport
(requires energy)
- Substances move in VESICLES
- Examples in cells:
 - GOLGI release packaged proteins
this way

Exocytosis

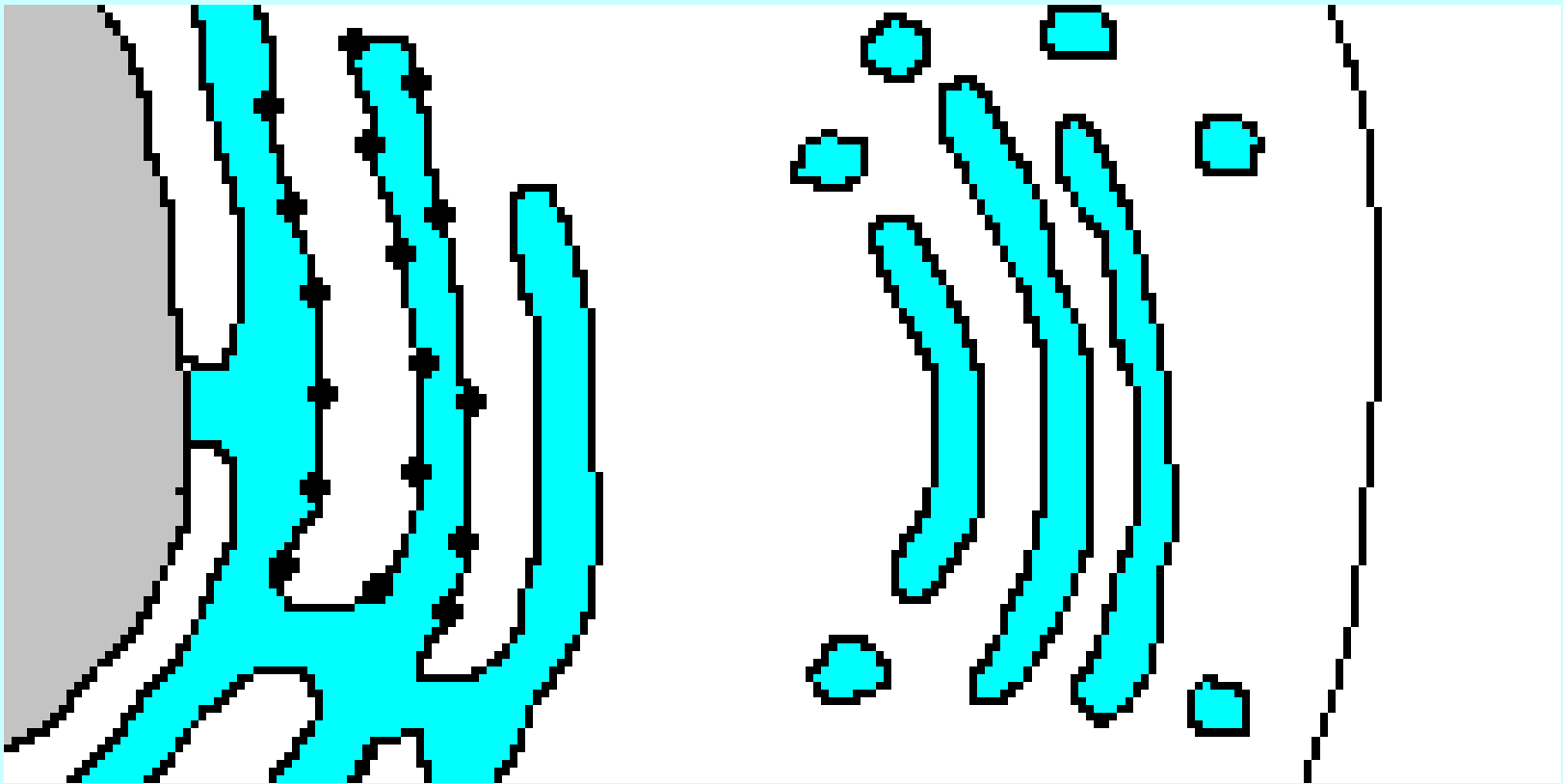


VESICLE



PLASMA
MEMBRANE

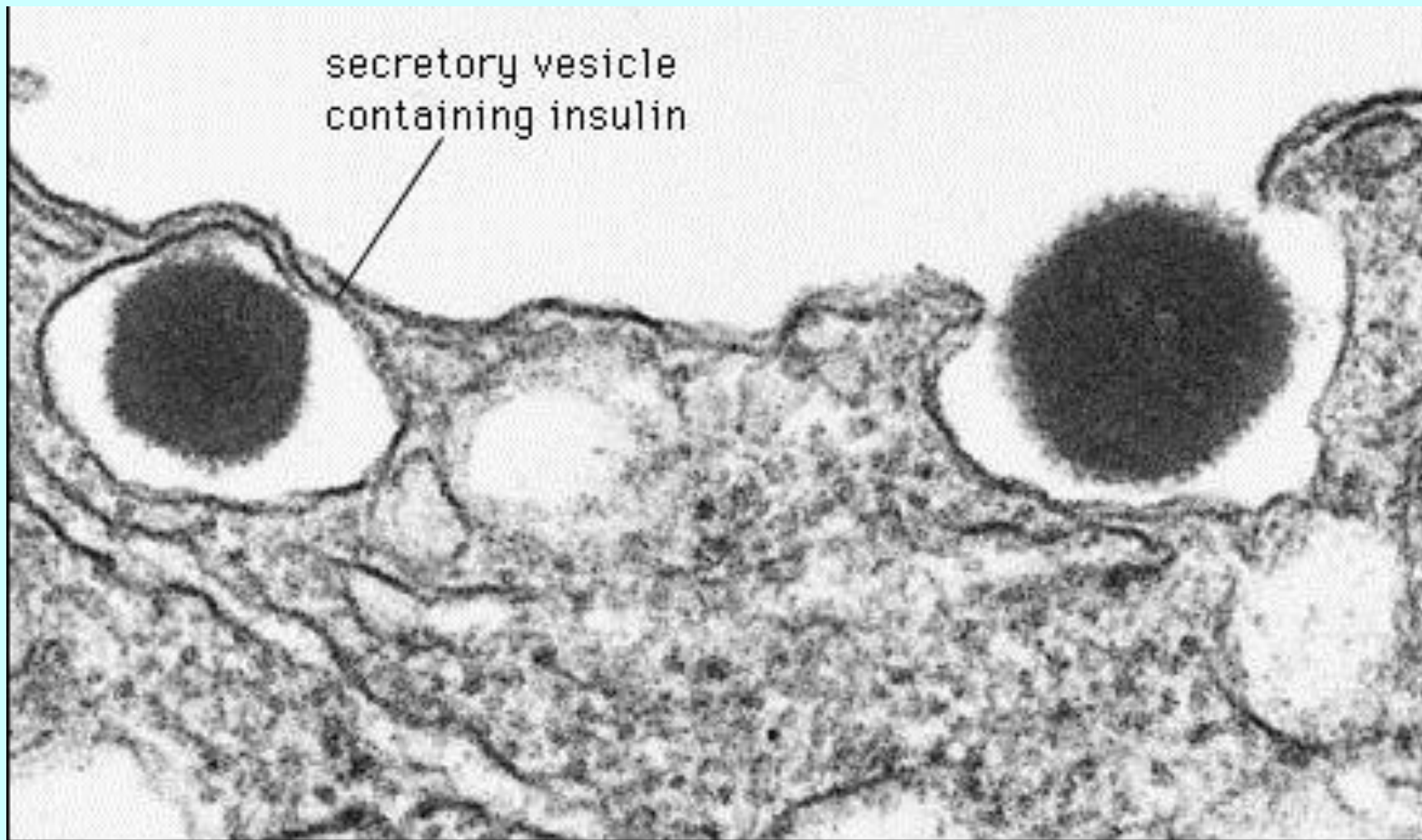
GOLGI BODIES USE EXOCYTOSIS



Animation from: <http://www.franklincollege.edu/bioweb/A&Pfiles/week04.html>

[See a Golgi movie](#)

INSULIN being released by pancreas cells using exocytosis

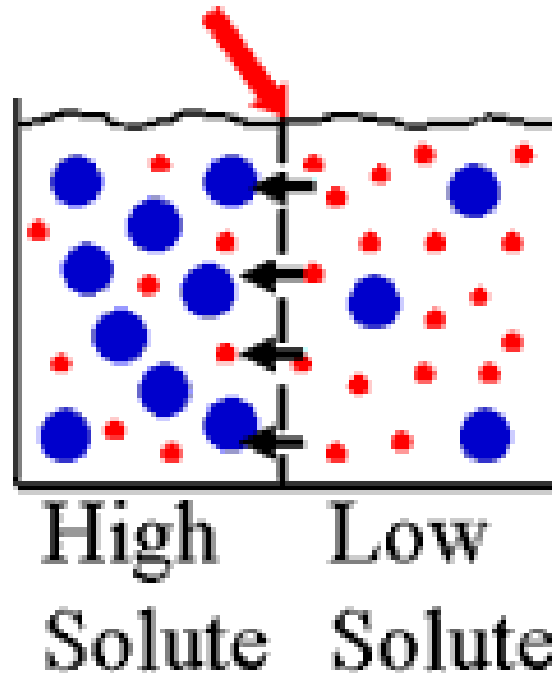


0.2 μm

<http://fig.cox.miami.edu/~cmallery/255/255ion/fig14x26.jpg>

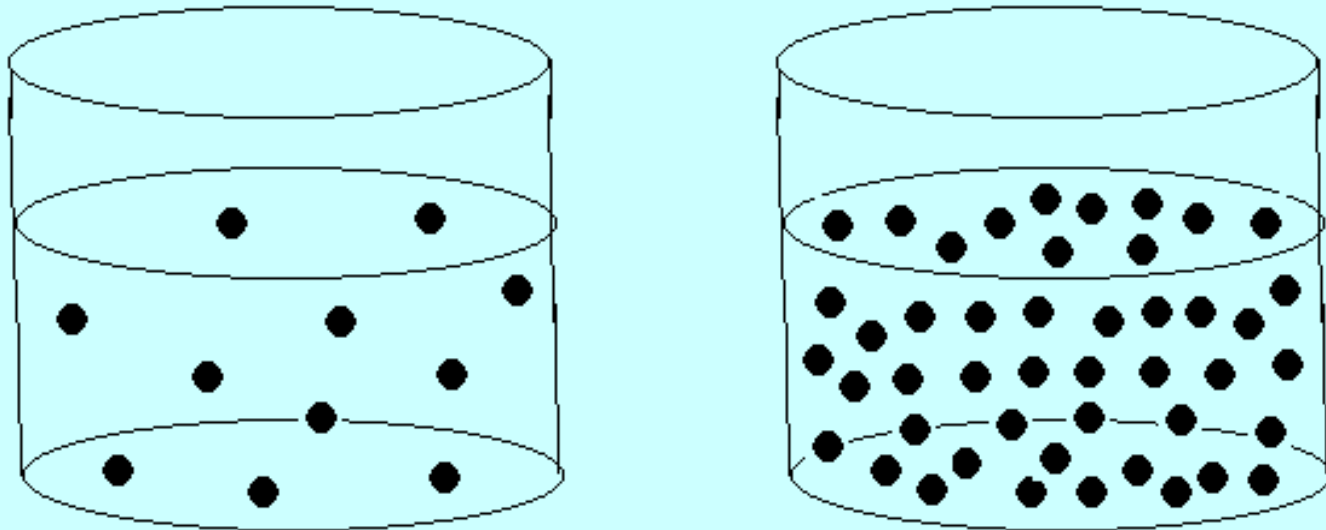
What if there is a difference in concentration but solute molecules can't move across a membrane?

Semipermeable membrane



WATER will move until concentration reaches equilibrium

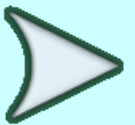
CONCENTRATION = mass of
a solute in a given volume of
solution



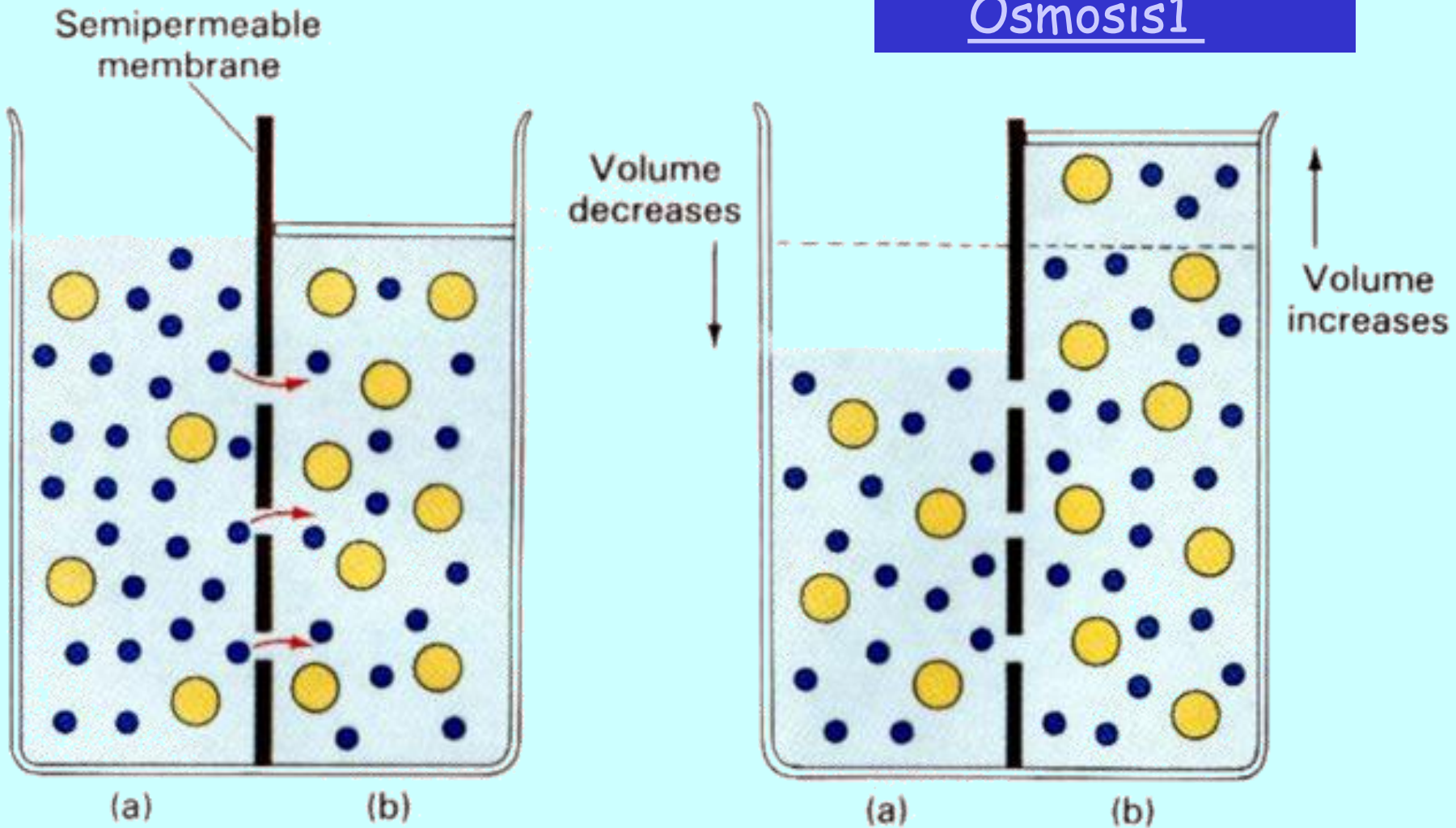
The MORE molecules there are in a given
volume the GREATER the concentration

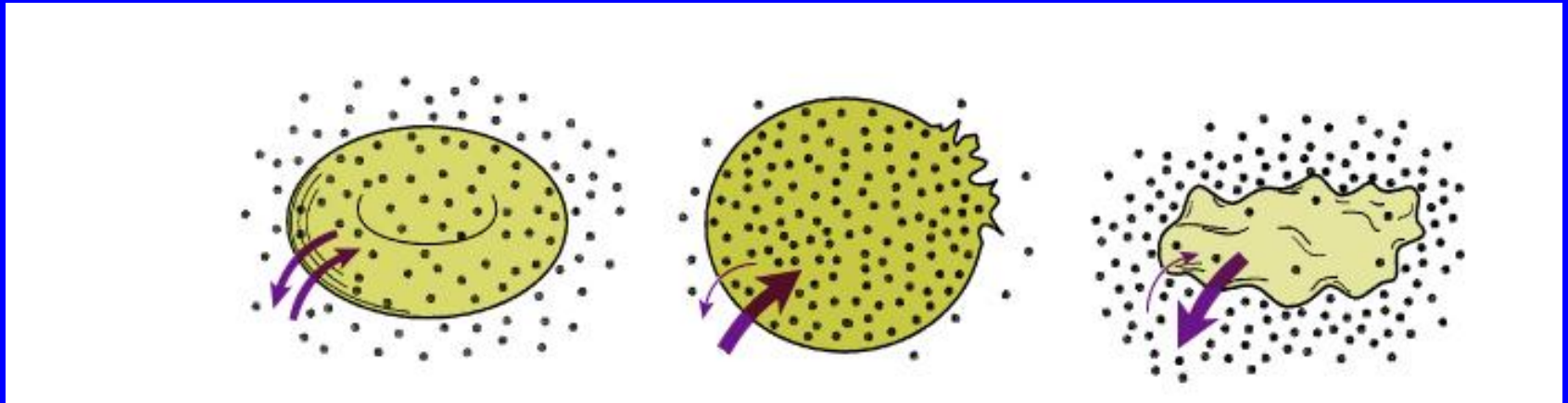


**See a video clip about
OSMOSIS -7B**

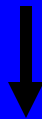


See an animation
Osmosis1



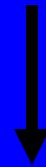


Isotonic Solution



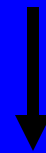
NO NET
MOVEMENT OF
 H_2O (equal amounts
entering & leaving)

Hypotonic
Solution



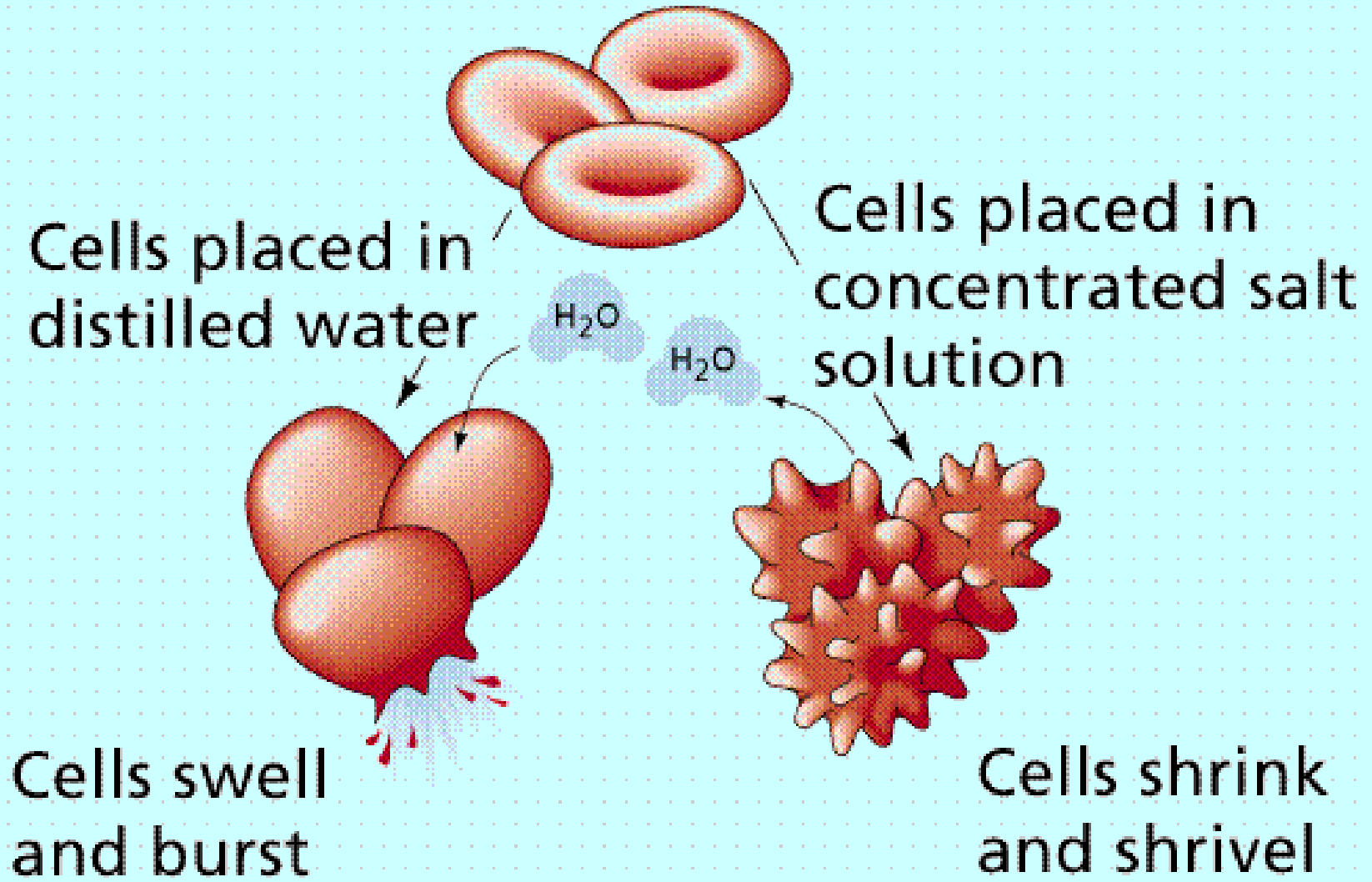
Cell Swells &
may burst

Hypertonic
Solution

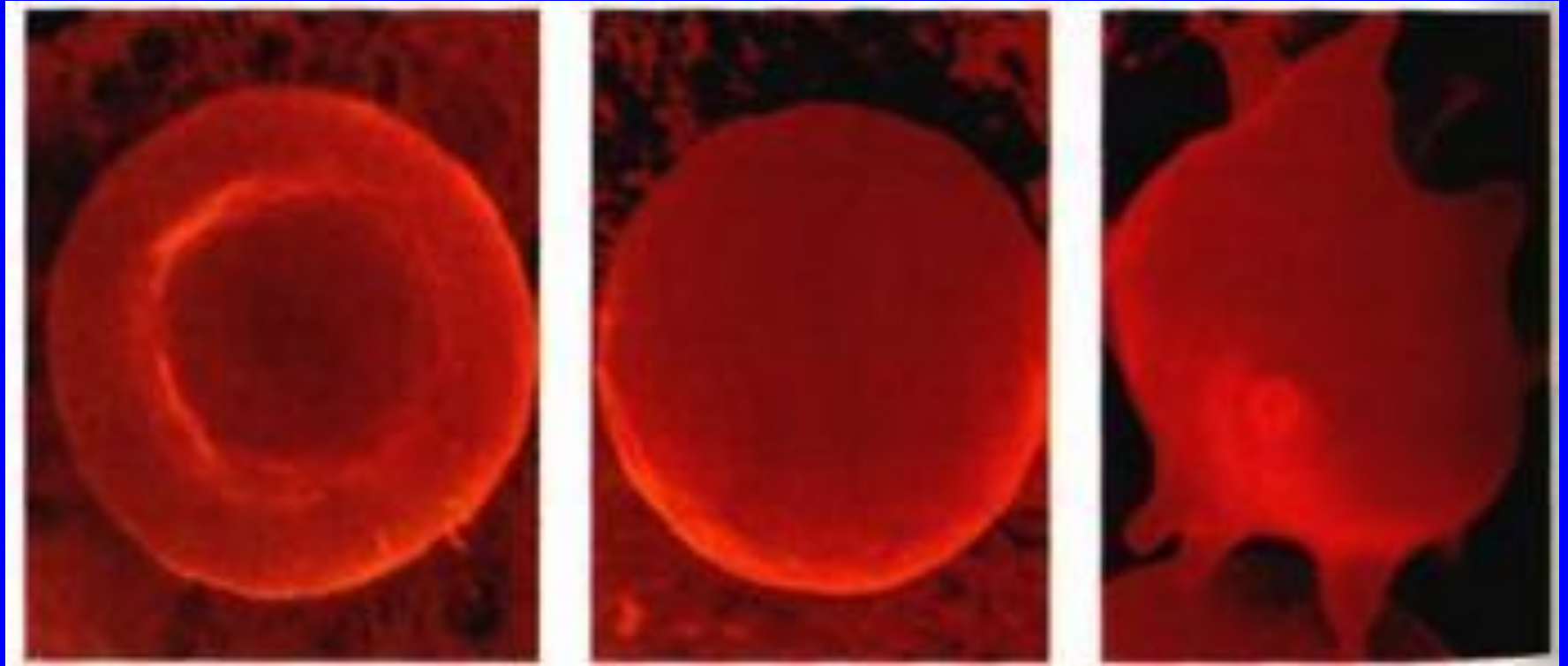


Cell shrivels up

Animal cells



Osmosis in Red Blood Cells

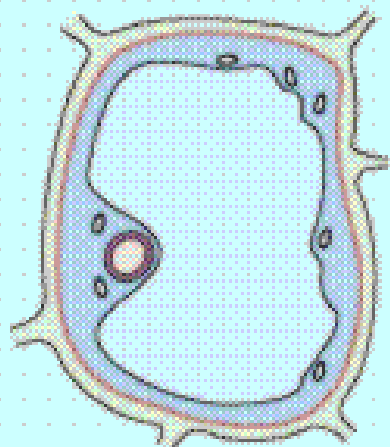
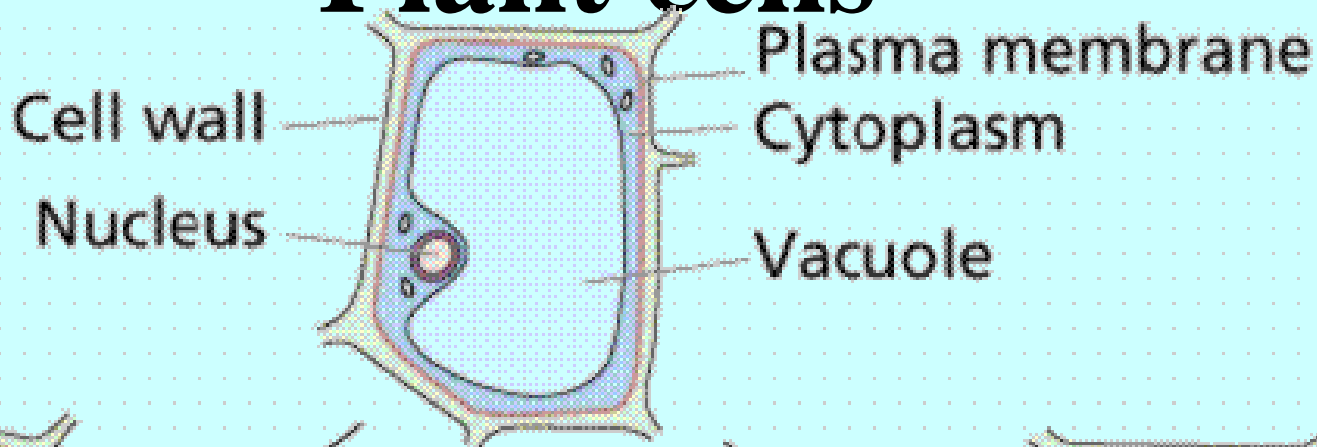


Isotonic

Hypotonic

Hypertonic

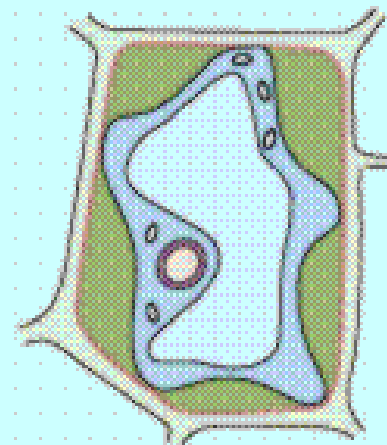
Plant cells



Plant cell placed in distilled water

Cell stiffens but generally retains shape

Plant cell placed in concentrated salt solution



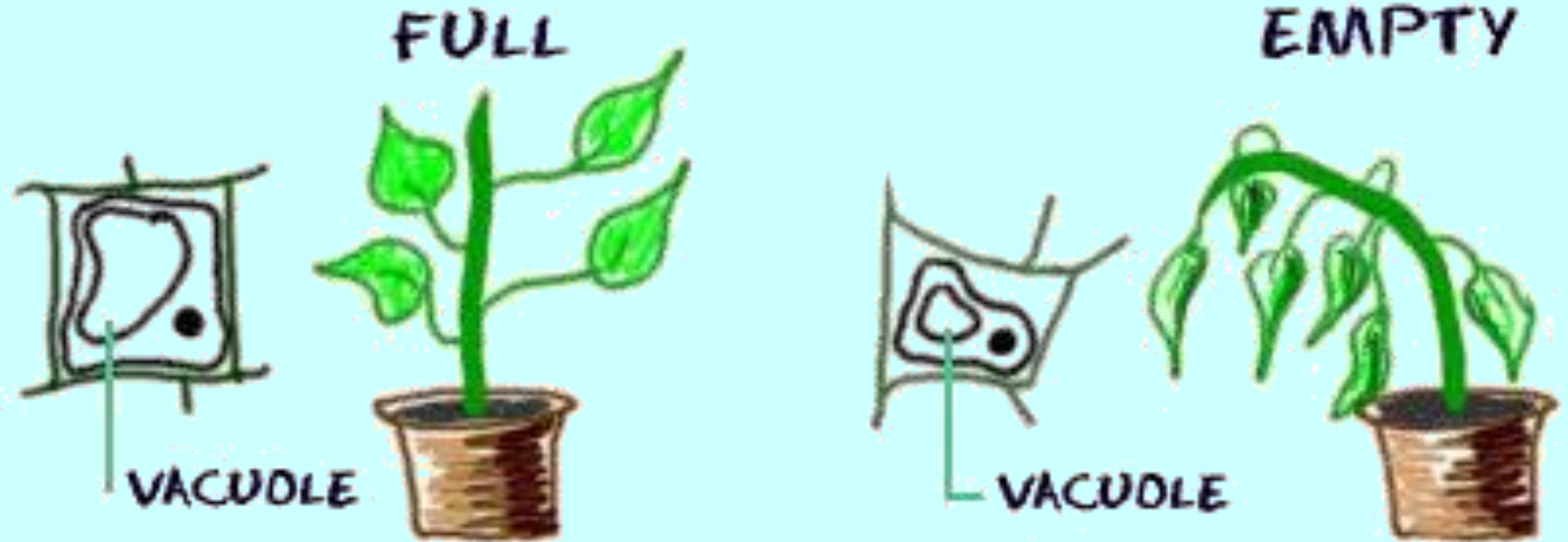
Cell body shrinks and pulls away from cell wall

CELL WALL

keeps

Plant cells from bursting

VACUOLES store WATER



http://www.biology4kids.com/files/cell_vacuole.html

OSMOTIC PRESSURE =

Pressure exerted by water during osmosis

SO WHAT?



Bath water is
hypotonic
compared to you

Sitting in the bathtub causes your fingers
and toes to wrinkle up when water
enters your skin cells by osmosis

Grocery stores spray water on their veggies to “plump them up”

